ISSN: 2249-894X Impact Factor: 5.2331(UIF)

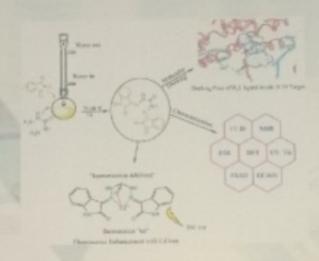
Volume - 7 | Issue - 9 | June - 2018

REVIEW OF RESEARCH

International Online Multidisciplinary Journal Journal No.: 4857h



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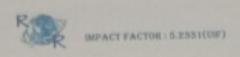
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Editor - In - Chief - Ashok Yakkaldevi



REVIEW OF RESEARCH

UGC APPROVED JOURNAL NO. 48514

255H: 2249-694X



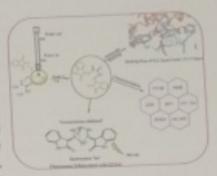
VOLUME - 7 | ESUE - 9 | JUNE - 2016

STUDIES OF SCHIFF BASE LIGANDS, ITS COMPLEXES WITH MERCURY (II) AND THEIR APPLICATIONS- A REVIEW

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ABSTRACT:

A variety of Schiff bases have been investigated. Mercury(II) with Schiff base ligands formed as supramolecular frameworks or multifunctional coordination polymers coordination geometry. Mercury (II) are also formed asymmetrical as well as nitrogen, sulpher and oxygen as donor Schiff-base complexes. Sometime mercury (II) are formed supramolecular complexes by non covalent linkages. The Schiff-base complexes are ability as biologically active substances, dyes, liquid crystals, luminophores and polymer stabilizers. The applications these Schiff-base complexes such as antimicrobial, antitumor, antidepressants, nematocide, antiphlogogistic and other medicinal significants. Another



applications in the paper industry and as preservative, fluorescent lamps, sensors and batteries. The Schiffbase are used to simultaneous detection and removal of mercury (II) ions from water.

KEYWORDS- Schiff base, supramolecular complexes, mercury(II), detection and removal.

The heavy and transition metal ions are very attractive because of their extremely toxic effect on INTRODUCTION the environment and biological systems. Among of these, mercury one of the-most poisonous and toxic pollutants that are serious to health problems. The mercury and its salts find wide applications in industrial processes and products as such paints, electrical equipments and batteries. The mercury has different forms in environment such as Hg⁰, Hg²⁺, and CH₃Hg⁴. The mercuric ion is neurotoxin as methyl mercury. The methyl mercury easily enter the environment and food chain of aquatic organisms and environment with mining, transportation and processes of mercury ores, dumping of industrial wastes into rivers and lakes, combustion of fossil fuels, pulp and paper industry, wood pulping, oil refining, solid waste burning, coal and gold mining, rubber processing, fertilizer industries, the use of mercury compounds as seed dressings in agriculture. 1,3,3 The inorganic mercury may be change into organic mercury like methyl mercury, which are more toxic than inorganic mercury. Methyl mercury triggers several serious problems for humans such as allergic reactions, brain and neurological damage. For determination and identification of mercury compounds are generally used optical chemical sensors (optodes). Essential steps in fabrication of optical sensor is immobilization of ionophores on transparent membranes. The ionophore immobilized are physical entrapment, sol-gel, multilayered films or chemical bonding methods⁴.

The combination of inorganic mercury with natural component in soils, sediments or water is forming the complexes, specially amino or sulfide groups, which are change the toxicity of respective ion⁵.

Review of Research

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ISSN NO:- 2249-894X

Impact Factor: 5.2331(UIF)

Vol.- 7, Issue - 9, June-2018



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